

Hierarchical Porous g-C₃N₄ as a Photocatalyst Nanomaterial

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Heterogeneous photocatalysts are in the spotlight recently due to the rise in environmental pollution and energy crisis. Graphitic carbon nitride (g-C₃N₄) material shows a potential photocatalyst with attractive properties such as metal-free, low bandgap, and high stability. However, the bulk g-C₃N₄ photocatalytic performance is insufficient to be work properly in practical applications. The hierarchical porous nanostructure is one modification that could be applied to improve photocatalytic activity. This review discusses the g-C₃N₄ material as a promising photocatalyst, including other potential modifications to improve its photocatalytic performance

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Introduction

The advancement of technology adoption in industry has led to an uplift in the standard of living. However, this cannot be secluded from environmental pollution. Hazardous chemicals used in the textile industry require its wastewater to be treated more sophisticatedly (Kiani *et al.*, 2020; Kuleyin, Gök, and Akbal, 2021). Likewise, the handling of hospital wastewater (Top *et al.*, 2020) and the coal chemical industry (Shi *et al.*, 2020). On the other hand, the environmental impact of mounting fossil fuel consumption has raised awareness to replace it with renewable and eco-friendly energy sources.

Photocatalysts emerge as a great approach to tackle those problems. The material can degrade the organic pollutants in the wastewater and produce hydrogen through light-induced water splitting (Chen *et al.*, 2014). Nevertheless, the performance of heterogeneous photocatalysts needs to be improved due to their shortcomings, e.g., low efficiency, stability, and selectivity. One of the optimizations is a fabrication of a hierarchical nanostructure (Li, Yu, and Jaroniec, 2020). Graphitic carbon nitride (g-C₃N₄) is a potential hierarchical photocatalyst due to its appealing properties such as stability, hardness, biocompatibility, and ease of modifying the surface (Dong *et al.*, 2014).

This mini-review describes the g-C₃N₄ material as a promising photocatalyst, including other potential modifications to improve its photocatalytic performance.

Graphitic Carbon Nitride g-C₃N₄

Atomic and electronic structure

Graphitic carbon nitride is a carbon-based layered polymer resembling graphene, which also consists of nitrogen. The material is a *p*-type semiconductor, having high chemical stability, high

photostability, and high activity under visible light irradiation (Masih, Ma, and Rohani, 2017). Among the seven known graphitic carbon nitride phases, g-C₃N₄ is considered suitable for visible sunlight activation due to its relatively low bandgap at 2.7 eV (Ajiboye, Kuvarega, and Onwudiwe, 2020).

The g-C₃N₄ structure consists of only carbon and nitrogen, which form the basic units, i.e., triazine (C₃N₃) and heptazine or tri-s-triazine (C₆N₇). Compared to triazine, the heptazine system exhibits lower energy; hence it is known as the most stable phase. Therefore, the g-C₃N₄ is mainly considered a conjugated π system linking a large number of tri-s-triazine rings based on van der Waals forces between layers (Zhang and Sun, 2018). The structure is shown in the following **Figure 1**.

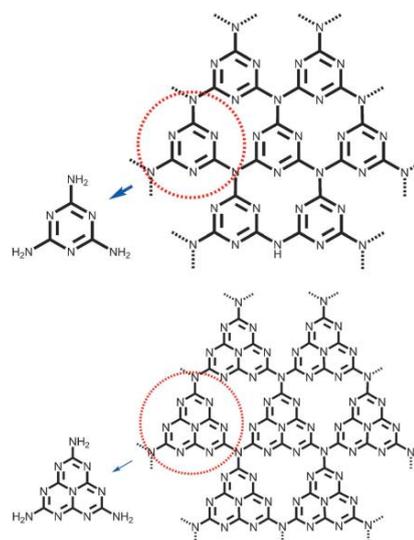


Figure 1. g-C₃N₄ structures: s-Triazine (top) and heptazine (bottom) (Wang, Wang and Antonietti, 2012).

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An ab initio study conducted by Wang *et al.* (2017) presumes that the heptazine-based $g\text{-C}_3\text{N}_4$ has two possible, stable arrangements, as shown in Figure 2.

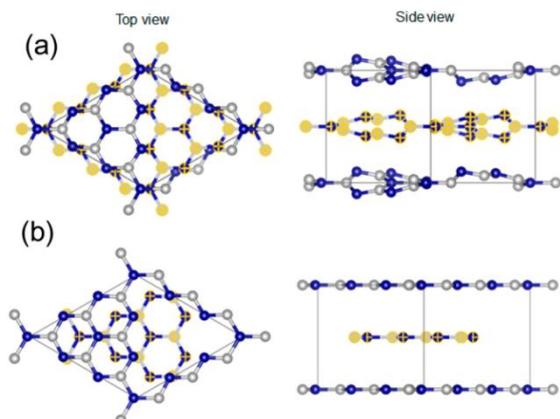


Figure 2. Predicted configuration of heptazine-based $g\text{-C}_3\text{N}_4$ (a) buckled; (b) planar (Wang *et al.*, 2017).

Their calculations showed that the buckled configuration has a direct bandgap of 2.87 eV. Meanwhile, the planar one has two bandgaps, direct 2.66 eV and indirect 2.27 eV. The experimental data reveal that the prepared material has a bandgap of 2.86 eV. This fact confirms that the heptazine $g\text{-C}_3\text{N}_4$ structure is likely to be in buckled configuration.

However, Jeong *et al.* (2019) found that $g\text{-C}_3\text{N}_4$ nanosheets have two distinct band gaps of 2.20 eV and 2.79 eV. These results correspond to planar heptazine $g\text{-C}_3\text{N}_4$ electronic structures. We may conclude that the bandgaps are tunable through morphology engineering.

Photocatalytic activity

In comparison to the conventional TiO_2 photocatalyst, $g\text{-C}_3\text{N}_4$ exhibits different photocatalytic behavior. While TiO_2 activity was enhanced by Pt addition, the graphitic carbon nitride showed an opposite effect, as shown in Figure 3.

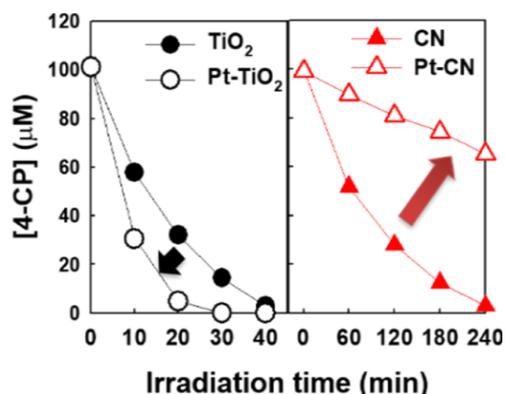


Figure 3. Photocatalytic degradation of organic substrate 4-chlorophenol (4-CP) by bare TiO_2 , bare $g\text{-C}_3\text{N}_4$ (CN), and their Pt-loaded counterpart (Lim *et al.*, 2020).

Lim *et al.* (2020) reveal that the contrary effects are due to hydroxyl radical formation in both photocatalytic systems. The adverse outcome of Pt on $g\text{-C}_3\text{N}_4$ is attributed to the fact that Pt breaks down the in situ generated H_2O_2 , which takes place as an intermediate of hydroxyl radical formation. Consequently, the hydroxyl radical formation is inhibited so that the photocatalytic activity decreased. The following Table 1 shows a brief comparison between $g\text{-C}_3\text{N}_4$ photocatalyst and TiO_2 .

Table 1. Comparisons between $g\text{-C}_3\text{N}_4$ and TiO_2

Properties	$g\text{-C}_3\text{N}_4$	TiO_2
Band-gap (eV)	2.7	3.2 (Dar <i>et al.</i> , 2014)
Working range	Visible light	Ultraviolet
Semiconductor type	<i>p</i>	<i>n</i>
Pore size (nm)	5 – 50 (Zhang <i>et al.</i> , 2018)	2 – 20, 25 – 150 (Tao <i>et al.</i> , 2011)

Synthesis and modifications

Bulk $g\text{-C}_3\text{N}_4$ can be synthesized by thermal condensation of assorted precursors with R-C-NH₂ units, as shown in Figure 4.

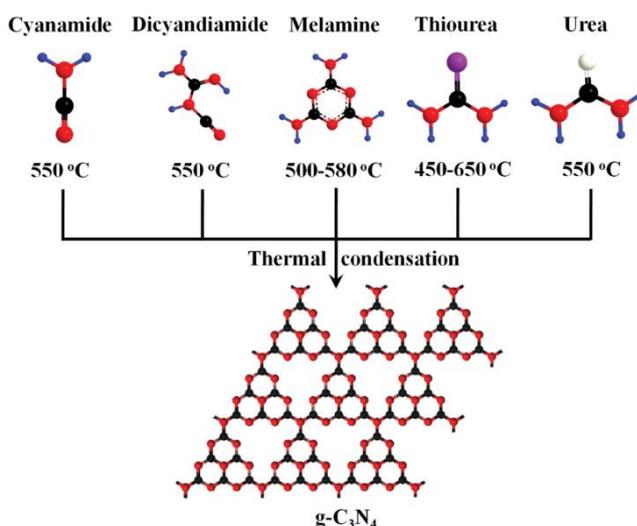


Figure 4. Preparations of $g\text{-C}_3\text{N}_4$ of various nitrogen-rich precursors (Cao *et al.*, 2015).

Nonetheless, the material prepared by this method has many defects, which led to a short visible light utilization range, low specific surface area, limited active site, and rapid recombination of the photoinduced carrier (Antil *et al.*, 2020; Zheng *et al.*, 2020). Hence, various efforts have been made in favor of improving the $g\text{-C}_3\text{N}_4$ photocatalytic performance.

Superstructure $\text{TiO}_2/\text{SiO}_2/g\text{-C}_3\text{N}_4$ hybrid has been reported to have an excellent performance to degrade berberine in seawater (Yu *et al.*, 2020). Azami *et al.* (2020) has successfully synthesized a $g\text{-C}_3\text{N}_4$ /fibrous titania silica photocatalyst. Oxygen doping has enhanced $g\text{-C}_3\text{N}_4$ for visible-light photodegradation of Rhodamin B (Tran *et al.*, 2021).

Hierarchical Porous g-C₃N₄

Hierarchical porous structure

Today, the fabrication of 3D hierarchical nanostructures has drawn significant attention due to their promising applications as photocatalysts (Liu *et al.*, 2015). The hierarchical structure comprises low-dimensional nanocrystal, e.g., nanosheets, nanowires, and nanorods, thus improving the material properties (Yang *et al.*, 2013). It is renowned that the hierarchical photocatalyst could present more porous sites as scattering the light to increase its absorption, a superior steric hindrance to inhibit the agglomeration of catalyst, and greater specific surface area to present more active sites (Li *et al.*, 2018; Wang *et al.*, 2019).

Synthesis of hierarchical porous g-C₃N₄

Hierarchical Sr-ZnO/g-C₃N₄ composite was prepared by diluting g-C₃N₄ powder via ultrasonic assistance. Then the solution was stirred in addition to Zn(NO₃)₂ hydrate, Sr(NO₃)₂, C₆H₉Na₃O₉, and NaOH solution. The obtained precipitate was rinsed with water and absolute ethanol consecutively and dried at 65°C for 15 h (D *et al.*, 2020).

On the other hand, (Liu *et al.*, 2020) successfully built the 3D hierarchical CdS/g-C₃N₄ two-dimensional heterojunction structure through facile hydrothermal method only with g-C₃N₄ powder, CdSO₄, and thiourea as precursors without any surfactants as assisting template. Lin, Yu, and Huang (2020) have successfully fabricated a cellulose-derived hierarchical g-C₃N₄/TiO₂ nanotube heterostructured nanocomposite by in situ coating thin g-C₃N₄ layers onto the surfaces of the TiO₂ nanotubes. The fabricated material is shown in Figure 5.

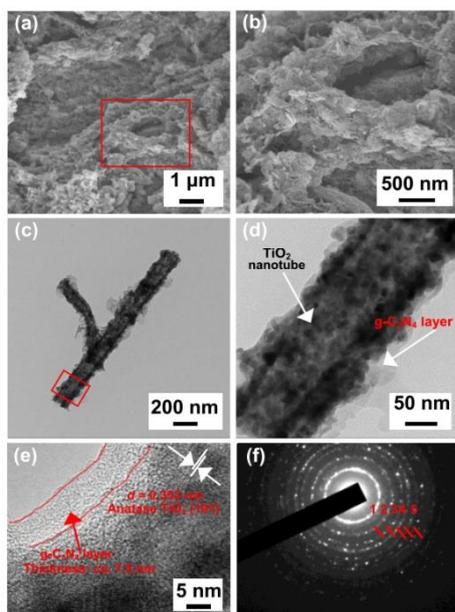


Figure 5. Electron microscopy images of the g-C₃N₄/TiO₂: (a, b) FE-SEM micrograph of the material, showing the hierarchical assemblies of the nanotubes. (c, d) TEM images of a single composite nanotube isolated from the assemblies. (e) HR-TEM micrograph of a part of the nanocomposite tube surface. (f) SAED pattern of the nanocomposite (Lin, Yu, and Huang, 2020).

Photocatalytic activities

The hierarchical Sr-ZnO/g-C₃N₄ and CdS/g-C₃N₄ show superior photocatalytic degradation towards organic pollutants than pristine g-C₃N₄. These findings are consistent with the previous discussion and prove that the modification is compatible with photocatalytic activity. While Lin, Yu, and Huang (2020) vary the g-C₃N₄ contents of 15%, 46%, and 66%. The comparison of g-C₃N₄/TiO₂ nanocomposite photocatalytic activity in visible light is shown in Figure 6.

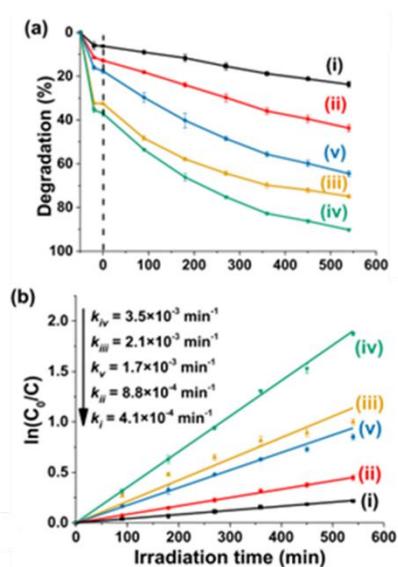


Figure 6. (a) Photocatalytic activity of g-C₃N₄/TiO₂ nanocomposite. (b) Respective pseudo-first-order simulations of the photocatalytic results by the referential TiO₂ nanotubes (i), referential g-C₃N₄ powder (ii), cellulose-derived 15% g-C₃N₄/TiO₂ (iii), cellulose-derived 46% g-C₃N₄/TiO₂ (iv), and cellulose-derived 66% g-C₃N₄/TiO₂ (v) nanocomposites (Lin, Yu, and Huang, 2020).

Summary

The g-C₃N₄ material is a promising photocatalyst due to its stability, environmentally friendly, low cost, and able to be activated by visible light. Despite the material yet having some shortcomings, a comprehensive option of modifications is available to be applied. The Hierarchical porous nanostructure is one of the modifications that has been proved to have more excellent photocatalytic activity than bare g-C₃N₄. Indeed, the material properties would improve if coupled with other modifications, such as being a composite or being doped with a particular element. The graphitic carbon nitride material commercial applications in the near future would create more exciting results.

Conflict of Interest

I have no conflict to declare.

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